

Course Name			-						
MASS AND ENERGY BALANCE Course Implementation, Hours/Week									
Code Seme		ster	Local Credit	ts	ECTS Credits	Theoretical		itorial	Laboratory
MET 248E	4		2		4	2		-	-
Department/Pr	ogram	Met	tallurgical and	Ma	terials Engineering	Department			
Course Type		Req	uired		Cours	se Language	Engl	lish	
Course Prerequ	isites	None							
Course Categor	y	Ba	sic Sciences	En	gineering Science	Engineering De	esign	Gener	al Education
by Content, %					80	20			
Course Description		Yield & Selectivity, Oxidation and Reduction, Description of Error, Precision, Accuracy and Repeatability, Measurement of weight, pressure, flow rate, Fundamentals of Material Balance Calculations, General Balance Equation, Procedure for Single-unit Process Material Balance Calculations, Balances on Multiple-unit Processes, Degree of Freedom Analysis, Recycling & By-pass Circuits, Balances on Continuous Steady-State Processes, Integral Balances on Batch, Semi-Batch and Continuous Processes, Systems with or without Chemical Reaction, Solution of sets of linear/non-linear equations: Introduction to computer aided tools for solving sets of linear equations, Energy Balances, Heat balance, Electrometallurgical Energy Balances, Simultaneous Material and Energy Balances, Examples of Materials and Energy Balances for Metallurgical Reactors							
Course Objectiv	This course covers the fundamental concepts in the field of Metallurgical Engineering, along with numerical examples from the existing industrial applications. Almost all processes, utilized in metal production technologies are covered within the framework of this course, which eventually outlines the background of the more technological courses offered in the following semesters.								
Course Learnin Outcomes					ith all desire s globally ar ning and lear n engineerir ocess for th g resources				
Textbook		Har	ndbook on Ma	teri	al and Energy Balai rris, Gordon Geiger				Processing,

Other References	 Basic Principles and Calculations in Chemical Engineering, David M. Himmelblau, Prentice Hall, 2004. Elementary Principles of Chemical Processes, Richard M. Felder and Ronald W Rousseau, 3rd Ed., John Wiley & Sons, Inc., 2000. Elements of Chemical Reaction Engineering, Fogler H.S., 4th Ed., Prentice- Hall, 2006. Mass and Energy Balances in Materials Engineering, M.E. Schlesinger, Prentice-Hall, 1996. 			
Homework & Projects	None			
Laboratory work	None			
Computer use	Working knowledge of MS Excel is helpful			
Other activities	None			
	Activities	Quantity	Effects on grading, 9	
	Midterm exam	2	40	
	Quiz	3	15	
	Homework	-	-	
Assessment Criteria	Project	-	-	
	Term Paper/Project	-	-	
	Laboratory Work	-	-	
	Other Activities	-	-	

Final exam

COURSE PLAN

1

45

Weeks	Topics	Course outcomes
1	Dimensions, System of Units, Conversion Factors	1
2	Stoichiometry, Balancing Chemical Reactions	1
3	Excess and Limiting Reactants, Yield & Selectivity, Oxidation and Reduction	1
4	Description of Error, Precision, Accuracy and Repeatability, Measurement of weight, pressure, flow rate	1, 5
5	Fundamentals of Material Balance Calculations, General Balance Equation	1, 5
6	Procedure for Single-unit Process Material Balance Calculations	1, 3
7	Material Balances; conservation of mass, mass balance analyses	1, 3
8	Balances on Multiple-unit Processes, Degree of Freedom Analysis	1, 3
9	Recycling & By-pass Circuits, Balances on Continuous Steady-State Processes	1-9
10	Integral Balances on Batch, Semi-Batch and Continuous Processes	1-9
11	Systems with or without Chemical Reaction, Solution of sets of linear / non-linear equations: Introduction to computer aided tools for solving sets of linear equations	1-9
12	Energy Balances, Heat balance, Electrometallurgical Energy Balances	1-9
13	Simultaneous Material and Energy Balances	1-9
14	Examples of Materials and Energy Balances for Metallurgical Reactors	1-9

Relationship between the Course and Metallurgical and Materials Engineering Curriculum

	Student Outcomes		Level of Contribution		
		1	2	3	
1	Ability to apply the knowledge of mathematics, science, and engineering principles to solve problems in metallurgical and materials engineering (ABET:a)			x	
2	Ability to characterize materials using standard and/or self-designed experimental methods and to evaluate the results (ABET:b)				



3	Ability to design a system or a process, taking into consideration of the desired specifications, quality, ethics and environment (ABET:c)		
4	Ability to communicate both orally and in the written form and to take part in, and provide leadership of the teams in the elucidation of engineering problems (ABET:d, g)		
5	Ability to define, formulate and solve engineering problems in the development, production, processing, protection and usage of engineering materials (ABET:e)		x
6	An understanding of professional and ethical responsibilities (ABET:f)		
7	An understanding of current/contemporary issues and impact of engineering solutions in broad cultural, national and global levels (ABET:h, j)		
8	A comprehension of the nature of engineering progress closely linked with the development of new materials and production processes. An ability to engage in life-long learning and a recognition of its necessity (ABET:i)		
9	Ability to use essential tools and techniques of modern engineering in the development, production, processing, protecting of the existing and new engineering materials (ABET:k)	х	

1: Little, 2: Partial, 3: Full

Course relationships with major elements of the field and material classes

		Level of Contribution		
		1	2	3
	STRUCTURE		Х	
	PROPERTIES			Х
	DESIGN EXPERIMENT/ANALYSE DATA		Х	
MAJOR ELEMENTS OF THE	PROCESSING		Х	
FIELD	COST/PERFORMANCE			
	QUALITY/ENVIRONMENT		Х	
	DESIGN PROCESS OR PRODUCT		Х	
	METAL			Х
	CERAMICS		Х	
MATERIAL CLASSES	POLYMERS			
	COMPOSITES			

1: Little, 2: Partial, 3: Full

Prepared by	Date	<u>Signature</u>
Prof.Dr. Cüneyt ARSLAN Doç.Dr. Cevat Bora DERİN	December 2017	